

Name: \_\_\_\_\_ Date: \_\_\_\_\_ Class Period: \_\_\_\_\_

## Algebra Final Exam STUDY GUIDE

*~Show Your Work!*

### PART 1

1. What is the degree of the monomial?  $7x^8y^6$
2. What is the degree of the polynomial?  $3a^4b^3 + 9a^2b^7$
3. Write the polynomial in standard form. Then name the polynomial based on its degree and number of terms.  $4x^2 + 2x^3 - 6 + 8x$
4. Simplify the sum.  $(2u^3 + 6u^2 + 3) + (2u^3 - 7u + 6)$
5. Find the difference.  $(-4x^2 - 3) - (3x^2 + 2x + 2)$
6. Simplify the expression  $x^8 \cdot 2y^{10} \cdot 5x^3$
7. Simplify the expression  $(7 \times 10^2)(7 \times 10^5)$ . Write your answer in scientific notation.
8. What is the simplified form of the expression  $(3k^3)^4$ ?
9. Write in simplified form.  $\frac{c^9d^{-3}}{c^{14}d^{-16}}$
10. Simplify the expression  $\left(\frac{m^{-1}m^3}{m^{-2}}\right)^{-3}$

# Algebra 2 Semester 2 Study Guide Answers

**United States. Division of Vocational  
and Technical Education**



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web he concluded that at a constant pressure the volume of the gas would expand at a particular rate for each degree of temperature rise that being 1 267 per degree in 1874 victor regnault obtained better experimental results showing this number to be 1 273 and concluded that the pressure would approach zero at 1 273 degrees c

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web the volume of 1 00 mol 1 00 mol of any gas at stp standard temperature 273 15 k and pressure 1 atm is measured to be 22 414l 22 414 l we can substitute 101 325kpa 101 325 kpa for pressure 22 414 l 22 414 l for volume and 273 15 k 273 15 k for temperature into the ideal gas equation and solve for r r

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web chemists sometimes make comparisons against a standard temperature and pressure stp for reporting properties of gases 273 15 k and 1 atm 101 325 kpa 1 at stp one mole of an ideal gas has a volume of about 22 4 l this is referred to as the standard molar volume figure 9 18

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web volume pressure kpa linear pressure kpa volume ml pressure kpa in part b we observed the relationship between temperature and pressure the pressure of the 125 ml flask was at 101 36 kpa when the temperature was at 23 2 c

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web pressure increases even though there were errors in the experiment the goal of the experiment was met the goal of declaring what type of relationship it is was clear using the data pressure kpa temperature c temperature k constant k p t 95 82 kpa 16 0 c 289 k 0 33 97 90 kpa 19 0 c 292 k 0 34 98 97 kpa 99 22 kpa 101 4 kpa 19 1 c

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surface and interfacial tension ph refractive index and vapor pressure is performed in the pvt laboratory pressure volume and temperature relationships of hydrocarbons are studied at the graduate level

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web the properties of water at constant volume can be represented as a function of pressure and temperature as shown in the diagram below figure 1 the pressure temperature relationship at constant volume the saturation point of water is the condition at which a phase change occurs from liquid to vapour or vapour to liquid

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web figure pageindex 5 the empirically determined relationships among pressure volume temperature and amount of a gas the thermometer and pressure gauge indicate the temperature and the pressure qualitatively the level in the flask indicates the volume and the number of particles in each flask indicates relative amounts

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