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Destiny Unchain Online

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～吸血鬼少女となって、
やがて「赤の魔王」と
呼ばれるようになりました～

MONTHLY SHONEN JUMP

Alexs Destiny Racing To Love Volume 4

K Payea



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water $n_a = 54 \text{ g} / 18 \text{ g mol}^{-1} = 3 \text{ mol}$ total number of moles $n_a + n_b = 0.5 + 3 = 3.5 \text{ mol}$ mole fraction of solute ethyl alcohol $x_b = n_b / n_a + n_b = 0.5 / 3.5 = 0.1429$

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40 g of H_2O molar mass of H_2O is 18 gm mole and molar mass of CH_3OH is 32 gm mole solution moles of $\text{CH}_3\text{OH} = 4.5 / 32 = 0.14$
mole moles of $\text{H}_2\text{O} = 40 / 18 = 2.2$ moles therefore according to the equation

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time the total pressure $p_a = x_a p_t$ mole fraction $x_a = \frac{\text{moles } a}{\text{moles total}}$ $\sum_{i=1}^n x_i = 1$ mole fraction of substance
 a is often denoted by x_a and

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web notice that the mole fraction has no units on it and is written as a decimal value do not change it to percent note of caution you could see the term mole percent it is simply the mole fraction multiplied by 100 for example in the problem just below the mole fraction of cinnamic acid is 0 2885 its mole percent would be 28 85

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