

Solutions to the Exercices of

C. J. Foot's Atomic Physics

Chenchao Zhao

Department of Physics, Beijing Normal University, Beijing, China

(Dated: June 28, 2011)

1 Early atomic physics

Key formulas:

$$\frac{1}{\lambda} = R \left(\frac{1}{n^2} - \frac{1}{n'^2} \right) \quad (1.1)$$

$$R_H = R \frac{m_H}{m_e + m_H} \quad (1.2)$$

$$a_0 = \frac{\hbar^2}{(Ze^2/4\pi\epsilon_0)m_e} = 0.529 \times 10^{-10} \text{ m} \quad (1.3)$$

$$E = - \frac{Ze^2/4\pi\epsilon_0}{2a_0} \frac{1}{n^2} \propto Z^2 \quad (1.4)$$

$$\Delta E = \frac{\alpha^2}{n^2} E \quad (1.5)$$

$$\alpha = \frac{e^2/4\pi\epsilon_0}{\hbar c} = 1/137 \quad (1.6)$$

$$\frac{f}{c} = R_\infty ((Z - \sigma_K)^2 - (Z - \sigma_L)^2) \quad (1.7)$$

$$\rho(\omega) = \frac{\hbar\omega^3}{\pi^2 c^3} \frac{1}{\exp \hbar\omega/kT - 1} \quad (1.8)$$

$$\frac{N_2}{g_2} = \frac{N_1}{g_1} \exp \left(- \frac{\hbar\omega}{kT} \right) \quad (1.9)$$

$$P = \frac{e^2 E_0^2 \omega^4}{12\pi\epsilon_0 c^3} \quad (1.10)$$

$$\Omega_L = \frac{eB}{2m_e} \quad (1.11)$$

$$\tau = \frac{6\pi\epsilon_0 m_e c^3}{e^2 \omega^3} \quad (1.12)$$

Atomic Physics Solution Manual Foot

CO Houle



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Table of Contents Atomic Physics Solution Manual Foot

1. Understanding the eBook Atomic Physics Solution Manual Foot
 - The Rise of Digital Reading Atomic Physics Solution Manual Foot
 - Advantages of eBooks Over Traditional Books
2. Identifying Atomic Physics Solution Manual Foot
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Atomic Physics Solution Manual Foot
 - User-Friendly Interface
4. Exploring eBook Recommendations from Atomic Physics Solution Manual Foot
 - Personalized Recommendations
 - Atomic Physics Solution Manual Foot User Reviews and Ratings

- Atomic Physics Solution Manual Foot and Bestseller Lists
- 5. Accessing Atomic Physics Solution Manual Foot Free and Paid eBooks
 - Atomic Physics Solution Manual Foot Public Domain eBooks
 - Atomic Physics Solution Manual Foot eBook Subscription Services
 - Atomic Physics Solution Manual Foot Budget-Friendly Options
- 6. Navigating Atomic Physics Solution Manual Foot eBook Formats
 - ePub, PDF, MOBI, and More
 - Atomic Physics Solution Manual Foot Compatibility with Devices
 - Atomic Physics Solution Manual Foot Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Atomic Physics Solution Manual Foot
 - Highlighting and Note-Taking Atomic Physics Solution Manual Foot
 - Interactive Elements Atomic Physics Solution Manual Foot
- 8. Staying Engaged with Atomic Physics Solution Manual Foot
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Atomic Physics Solution Manual Foot
- 9. Balancing eBooks and Physical Books Atomic Physics Solution Manual Foot
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Atomic Physics Solution Manual Foot
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Atomic Physics Solution Manual Foot
 - Setting Reading Goals Atomic Physics Solution Manual Foot
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Atomic Physics Solution Manual Foot
 - Fact-Checking eBook Content of Atomic Physics Solution Manual Foot
 - Distinguishing Credible Sources

13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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